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## Nucleosides, Nucleotides and Nucleic Acids

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### A Computer Generated Model of Adenosine Receptors Rationalising Binding and Selectivity of Receptor Ligands

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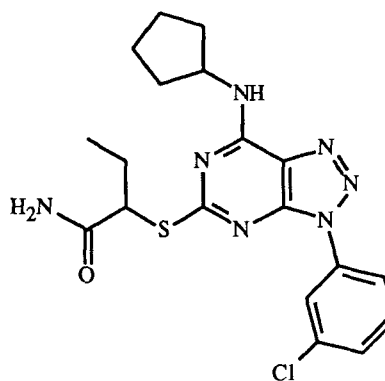
A COMPUTER GENERATED MODEL OF ADENOSINE RECEPTORS  
RATIONALISING BINDING AND SELECTIVITY OF RECEPTOR LIGANDS.

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**Abstract** Computer graphic analyses on a broad spectrum of adenosine receptor ligands has shown that both the A1 and A2 adenosine receptors have three binding sites. The spatial relationship of these three binding sites has been defined. Adenosine orientation at A1 and A2 is different.

Over the last several years we have been exploring the role of the heterocyclic rings in affecting affinity to adenosine receptors by synthesis and radioligand binding assay of a variety of adenosine analogues where the heterocyclic ring has been modified. In particular our work, in addition to purine systems, has involved pyrazolo[3,4-*d*]pyrimidines, triazolo[4,5-*d*]pyrimidines and thiadiazolo[4,5-*d*]pyrimidines. During the course of this work we observed a lack of additivity in the substituents, typified by 3-(3-chlorophenyl)-7-(N-cyclopentylamino)-5-(2-butanamidyl-

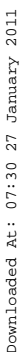


(1)

thio)[1,2,3]triazolo[4,5-*d*]pyrimidine (1), which showed a decreased affinity for the A1 receptor compared to analogous mono-substituted compounds. At the same time, similar observations were reported on substituting adenosine at both the C2 and N<sup>6</sup> positions. For example, 2-(phenylamino) substitution of the A2-selective agonist N<sup>6</sup>-[2-(3,5-

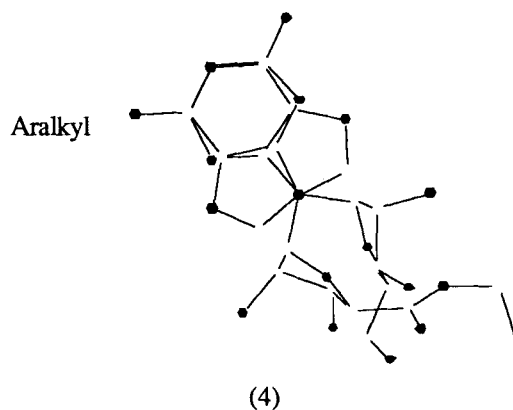
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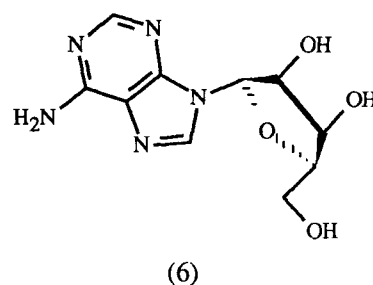
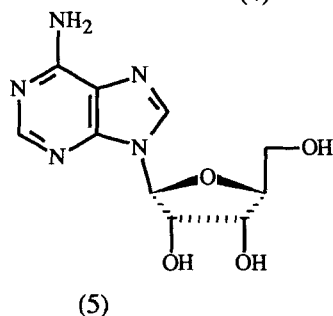


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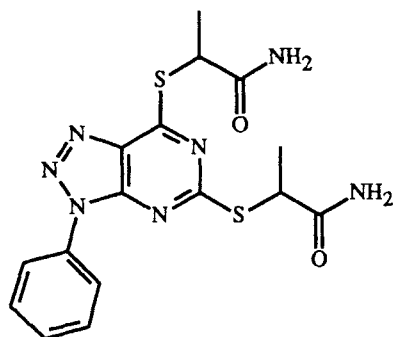
and (6) thereby allowing determination of how adenosine binds at A1 and A2 receptors. The orientation of adenosine in (5) can undergo a  $180^\circ$  rotation about the 4,5-bond, so that the ring oxygen of the ribose is now into the page and a rotation of the purine ring through about  $130^\circ$  anti-clockwise would give adenosine oriented as in (6).



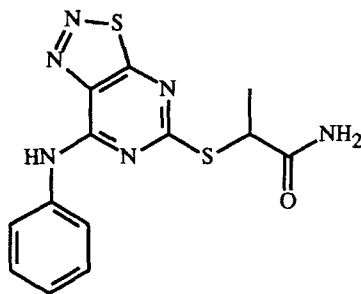
In (4) the 2' hydroxyls are  $5.88\text{\AA}$  apart, the 3' hydroxyls are  $5.54\text{\AA}$  apart



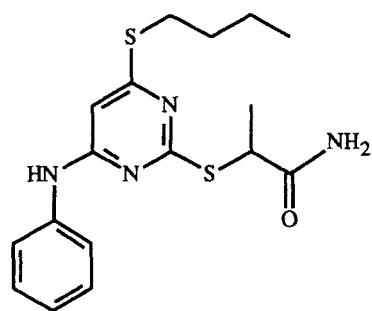
This model is further substantiated by examples (7), (8), (9) from a series of heterocyclic compounds which have the same order of A1 binding ( $IC_{50}$ 's around  $10^{-5}M$ ) such that when the phenyl ring occupies the hydrophobic domain, the other chains occupy a common region, confirming that it is the relationship of the phenyl and amide that is important in binding. Fitting of these compounds against NECA has shown that both amides are in close proximity. (10) shows the fitting of the ribose of NECA to the amide side chains where the O - O' distance was  $1.0074\text{\AA}$ , the C - C' distance was  $1.0443\text{\AA}$  and the N - N' distance was  $0.8917\text{\AA}$ . Consistent with the six-membered ring of the heterocycle being conserved, (9) has a similar affinity for the A1 receptor. Lack of additivity has been reported with N<sup>6</sup>,5' doubly modified agonists.<sup>3,4</sup> N<sup>6</sup> substitution forces the ribose into the A1 ribose binding pocket so that NECA type substitution in the ribose would be expected to lower affinity.



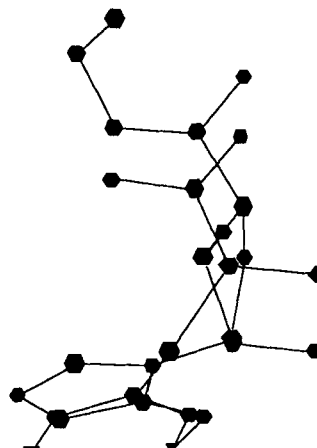
(7)



(8)



(9)



(10)

Adenosine receptors have three binding pockets:- a hydrophobic binding domain, a central aromatic binding domain and a ribose binding domain. The spatial relationship of these three binding sites has been defined. The different orientation of adenosine at the A1 and A2 receptors could be explained by a hydrogen bonding interaction between the respective receptor proteins and a N<sup>6</sup> hydrogen.

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